

## **Around 1990**

### **Usage of shallow junction and silicide source/drain techniques**

#### **Process Technology ~**

The CMOS transistor has continued miniaturization following the scaling rule and has achieved high performance. However, as it entered the 0.35 $\mu\text{m}$  generation, performance improvement came to be limited even when scaling was done. One of the causes was an increase in the parasitic resistance of the gate electrode and the source/drain diffusion layers. In order to solve this problem, SALICIDE (Self-Aligned siLICIDE) technology came to be used.

Salicide is a technique of forming metal silicide on silicon or polysilicon surface in a self-aligned manner. After forming transistors, a metal is deposited and heat treatment is applied to make it selectively react with the exposed Si surface to form a metal silicide. The metal film on the insulating film is then removed and a self-aligned metal silicide is formed. Historically it was not new, but its practical application started from 0.35 $\mu\text{m}$  CMOS generation.

It was first applied to TiSi<sub>2</sub>, titanium (Ti) and Si alloy. Since the diffusing element is Si, Si diffuses also into Ti on the insulating film, forming TiSi<sub>2</sub> after prolonged heat treatment. In order to avoid this, heat treatment at a high temperature for a short period of time, that is, RTP (Rapid Thermal Process) technology, was necessary for the alloy formation.

As the miniaturization progressed, the phenomenon called a thin line effect occurred in which TiSi<sub>2</sub> stopped transition to a low resistance phase. Although a high temperature heat treatment was necessary for the phase transition, agglomeration occurred when a high temperature process was used, and there arose a problem that the appropriate process temperature range for the stable formation of TiSi<sub>2</sub> was extremely narrow. Although the problem could be alleviated by making the Si surface amorphous, from the 0.18 $\mu\text{m}$  generation onwards, CoSi<sub>2</sub> alloy of cobalt (Co) and Si which did not have these problems came to be used.

However, there is also a problem with CoSi<sub>2</sub>. In the reaction between Co and Si, since the diffusion species is Co, Co easily penetrates into Si in a spike shape. It breaks through the shallow junction and leakage current increases in the further miniaturization of processes. As a result, NiSi is adopted in 0.1 $\mu\text{m}$  and beyond. As described above, advancement of salicide technology is indispensable for realizing further miniaturization and high performance.